

	Search Text
1	"5021999"
2	((("5021999") or ("5027171") or ("5111430") or ("5253196") or ("5293560") or ("5317535") or ("5388069") or ("5424993") or ("5430670") or ("5434815") or ("5438544") or ("5467306") or ("5477485") or ("5485422") or ("5493140") or ("5508543") or ("5627781") or ("5670790") or ("5714766") or ("5754477"))).PN.
3	((("5021999") or ("5027171") or ("5111430") or ("5253196") or ("5293560") or ("5317535") or ("5388069") or ("5424993") or ("5430670") or ("5434815") or ("5438544") or ("5467306") or ("5477485") or ("5485422") or ("5493140") or ("5508543") or ("5627781") or ("5670790") or ("5714766") or ("5754477")).PN.) and (nano near2 crystal\$8)
4	("438/452").CCLS.
5	((("438/452").CCLS.) and (oxid\$8 near7 temperature\$2)

threshold

voltages, it may be necessary to tolerate more leakage current than desired or

accept less area efficiency. Several solutions have been identified to

minimize such trade-offs. For example, field oxide growth under High Pressure

Oxidation can limit dopant diffusion by reducing the oxide growth temperature

while sustaining a high rate of oxidation; and a combined germanium-boron

implant favorably reduces the boron diffusion rate, thereby reducing the loss of

boron by diffusion into the oxide and by lateral diffusion. It is also well

known that with a chlorine implant the oxidation rate can be increased and the

time required for oxide growth shortened. That is, by growing the field oxide

faster there is less time for a highly mobile dopant species such as boron to

diffuse into the oxide and, overall, a lower implant dose can be used to create

the channel stop. Lower implant doses result in less lattice damage.

DEPR:

According to the invention it is now recognized that satisfactory net oxide

growth rates suitable for a high volume, cost sensitive, manufacturing

environment, can be sustained while process conditions are sequentially altered

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BSPR:

Numerous problems have been reported in cases where channel stop implants are used in combination with field oxides. See Wolf, Silicon Processing for the VLSI Era, Volume 21, Chapter 2 for a general discussion. Specifically, lattice dislocations generated during channel stop implantation are known to result in stacking faults during subsequent thermal processing such as the aforementioned field oxide growth. When these oxidation induced stacking faults (OISFs) extend into the active regions they can cause leakage currents which degrade device performance. In the past this has meant that certain performance specifications could only be met at the expense of others. Thus, in order to design an isolation structure with minimally sufficient parasitic

to inhibit OISF's. This is to be distinguished from shrinkage of stacking faults, which has been observed at high temperature, long oxidation conditions.

(See Lin, et al., J. Electrochem. Soc.: Solid-State Science And Technology,

May 1981.) That is, a sequence of process steps is provided which (1) assures

relatively short oxidation cycles suitable for the manufacturing environment,

(2) provides acceptable levels of dopant outdiffusion, and (3) inhibits OISF

development in the first instance.

DEPR:

The atmosphere is then modified to begin growth of field oxide under high

temperature, low oxidation rate conditions. For example, start at 6 percent

oxygen, 94 percent nitrogen for approximately 30 minutes, followed by 30

percent oxygen, 70 percent nitrogen for about 15 minutes. To the extent not

all of the nucleation sites have been removed during the preceeding anneal

step, application of high temperature, low growth rate conditions during

initial stages of field oxide growth will minimize creation of OISFs.

Variants

of these conditions, such as a single, but longer, growth at 10 to 20 percent

oxygen content may also provide satisfactory results.

DEPR:

An important feature of the invention runs counter to conventional practice of growing field oxide under conditions which result in undesirable diffusion of dopant species combined with a high oxidation rate. Instead, initial growth of field oxide occurs under high temperature, slow oxidation rate conditions to minimize development of OISF's. Further, according to the invention, accompanying concerns relating to loss of dopant species by high temperature diffusion during the slow oxidation process are mitigated by the aforementioned anneal which drives the dopant distribution away from the oxide 44 and deeper into the device layer. The structure 10 with growth of the minor portion 56 of field oxide, e.g., 200 Angstroms or more, is shown in FIG. 1j. It is characterized by a highly efficient channel stop implant and a low defect density in portions of the device layer 20 beneath the field oxide.

DEPR:

The structure 10 is next subjected to a higher oxidation rate to complete the field oxide growth. The majority of field oxide growth proceeds under high temperature conditions, e.g., 1050 C., for 180 minutes in steam. The temperature is then gradually decreased (3 degrees C. per

minute) to 700 C. and at that point the wafer structure having field oxide formations 62, commonly one micron thick, is pulled from the furnace. See FIG. 1k. Experimental data indicates that performing the rapid field oxide growth after growth under the high temperature, slow oxidation rate conditions does not adversely affect the defect density of the device layer. Specifically, the density of OISFs under the field oxide formations 62, in the region of p-type channel stop implants, has been measured as low as 5 per cm<sup>2</sup> through the device layer 20. This is to be compared to OISF densities observed in conventionally processed field oxides exceeding 1E6/cm<sup>2</sup>.

#### DEPR:

A process has been described which inhibits OISFs and yet provides net oxide growth rates suitable for a high volume, cost sensitive, manufacturing environment. Numerous variations of the invention apart from the disclosed embodiments will be apparent. For example, the aforescribed anneal to be performed in the nitrogen atmosphere at 1050 C. for 75 minutes need not occur in an inert environment. The anneal could also be performed in an oxidizing ambient that shrinks OISFs, such as a high temperature, low